

# Flame retardant smoke suppressant coatings for PVC sheathed electrical cables

Sunil K Sharma & N.K. Saxena  
 Scientist, Fire Research  
 Central Building Research Institute  
 Roorkee 247 667

## ABSTRACT

*Fire Spread along electrical cables may be mitigated through application of flame retardant Smoke suppressant (FRSS) intumescent coatings as fire protective layers to electrical cables as well as by using FRSS additives based PVC compositions for sheathing of electrical cables. Metal based organic (MBO) complexes of  $Mo^{+6}$  and  $Cr^{+3}$  were used as FRSS in the present study. FRSS intumescent coatings were applied on cables and their fire performance was evaluated using national and international standard procedures. The coated cables did not show any surface spread of flame on exposure and the generation of smoke was also found to be very low. Plasticized PVC compositions were prepared and found to have excellent smoke suppression. The limiting oxygen index was found to be fairly high both for the PVC compositions and the intumescent coatings developed.*

## Introduction

**F**IRE is an unexpected event, which is often frightening as well as dramatic. It can lead to loss of life and expensive damage to property and equipment. Large quantities of cables are now being used because of the modern trends in construction, growing need for electronic communication as well as for industrial applications. Modern production facilities also demand long runs of power and control cables all over the plant area. Any fire incidence in the vicinity of the cables may result in the initiation of fire at the cable surface and flame spread along it. Fortunately PVC, the material generally used for electrical power cable sheathing, is among the safer materials in a fire scenario basically because of its chlorine contents that renders it self-extinguishing. When exposed to heat

PVC emits hydrogen chloride into the gaseous phase and forms a cross-linked char in the solid phase. HCl reacts with the energy rich, oxygen-based, chain branching free radicals in the gaseous phase which propagate the flame, and thus acts as a chain terminator. The char on the other hand insulates parts of the solid phase; more remote from the fire source; from heat spread as well as reduces oxygen penetration. PVC has been combined with a variety of additives to improve upon the performance of several properties necessary for wires and cable sheathings however, many of these additives result in a compromise on its fire performance. Rapid and unrestrained propagation of fire can be checked either by addition of flame retardant smoke suppressant (FRSS) additives during preparation of PVC sheets or by application of an effective fire retardant coating on PVC

sheathed cables. Many FRSS for PVC have been reported literature<sup>1-3</sup>. These can be carbon fillers, additives and intumescent coatings. Essentially all of the smoke suppressant additives are either metals or metal compounds and are used only in small quantities. Fillers, on the other hand, are used in large quantities and may result in unfavorable mechanical properties. The FRSS additives can be used during polymer processing, or they can be applied to the existing cables.

There are two types of coatings which retard the spread of fire. The fire retardant non-intumescent coating, uses additives such as boric acid, antimony trioxide and chlorinated compounds. The intumescent coating does not support combustion<sup>1,4-6</sup>

**TABLE 1: COMPOSITIONS OF FIRE RETARDANT INTUMESCENT COATINGS**

Ingredients	Parts by weight for formulation numbers			
	1	2	3	4
Phosphate	14.0	12.0	15.5	17.0
Amide	12.0	10.0	9.0	11.0
Polyol	10.0	12.0	14.0	8.0
Pigment	5.0	3.0	6.0	4.0
Thickener	0.30	0.30	0.24	0.24
Filler	8.0	9.0	6.0	10.0
Co-polymer emulsion	33.0	35.0	33.0	33.0
Molybdenum complex	2.0	-	-	2.0
Chromium complex	-	2.5	2.0	-
Antifoaming, wetting & antisetling agent	0.20	0.20	0.26	0.16
Water	15.5	16.0	14.0	14.6

type is called fire retardant intumescent coating, which, on heating produce residues, which are swelled by escaping gases. A combustion residue is puffed up in order to produce a tough insulating foam over the surfaces to protect the materials<sup>1,7-11</sup>. These coatings perform better than the non-intumescent ones. Efficacy of fire retardant coatings in improving the burning behaviour of wood and wood based products has been studied by many researchers<sup>1,4-10</sup>. However, sufficient attention has not been paid towards use of suitable coatings to impart fire protection to electric cables.

This paper deals with the fire retardant smoke suppressant metal based organic (MBO) complexes used as FRSS additives in compositions for PVC sheets and in intumescent coatings that provide fire protection to PVC sheathed electrical cables.

## Experimental procedure

### Preparation of FRSS coatings

A few formulations of flame-retardant smoke suppressant intumescent coatings were prepared by using different types of phosphate amides, polyols, propanediono complex of molybdenum/chromium and binder in different weight ratios. Their compositions are given in

Table 1. The coatings were prepared by mixing fire retardant smoke suppressant ingredients of 325-400-mesh size with 2-5 percent solution of thickener along with appropriate quantity of antisetling, wetting and anti-foaming agents. Vinyl acetate and vinyl versatate copolymer emulsion (binder) was modified by reacting with dihydroxydimethylol ethylene urea (DHDMEU) and polymeric plasticiser to increase the adhesion to the PVC as well as to improve flexibility of the coating. The

required quantity of this modified binder was added to the above contents to make homogenous mixture in order to obtain brush consistency by adding enough water. It was stirred vigorously with a heavy-duty stirrer for one hour. Different thicknesses of FRSS intumescent coating thus prepared were applied with brush on PVC insulated electrical cables.

### Preparation of PVC specimens

Pentanedione complexes of molybdenum and chromium (FRSS additives) were synthesized using 2,4 pentanedione (acetylacetone)<sup>12</sup> by replacing the hydrogen of the hydroxyl group of the enol form of the 1,3 diketone by metal thus resulting in the formation of a chelate ring. PVC specimens were prepared by mixing virgin resin with different ingredients, including heat stabilizers

and lubricants such as dibutyl tin dilaurate. Ingredients were used in quantities depending on experiments (Table 1). The coatings were mixed thoroughly with PVC. Plasticizers such as lub and plasticizers were used and the mixing continued until the consistency of the mixture was the form of a free flowing dry mix thus obtained. The mixture was heated on an electrically heated plate at 140-150°C for 5 to 10 minutes. The coating was applied on the formulation was molded into sheets of 100 x 150 mm using a compression machine at 175°C, followed by cooling. Since the modulus of MBO additive was low, the smoke emission is comparatively small concentration decided to use pentanedione of molybdenum (VI) at a concentration of up to 3 parts per hundred (phr) loading. Samples were prepared using both the phthalate (DOP) and the dibutyl phosphite - TCP as plasticizers.

## Performance evaluation

The fire retardant coatings prepared, as above, were tested on PVC insulated electrical cables. The performance of the coatings was studied in terms of flame propagation and char

**TABLE 2: RECIPE AND PROCESSING CONDITIONS FOR PLASTICIZED POLY (VINYL CHLORIDE) SHEETS**

Ingredients	Control sample
Poly (vinyl chloride) resin	100 phr
Plasticizer	30-90 phr
Processing aids	2 phr
FRSS	0 phr

### Processing conditions

Dry blending at room temperature for 20 minutes  
Mastication on two-roll mill at 165°C for 10 minutes  
Compression molding at 175°C

\* Phr: parts per hundred parts of resin

# TINTIUM

## HIGH PERFORMANCE

# YELLOW ORGANIC PIGMENTS

*Will be the only organic yellow pigments  
in future*

- ▷ for fade resistant automotive and quality paints
- ▷ for powder coatings and plastics
- ▷ migration free
- ▷ tinctorially strong
- ▷ reasonably priced

*From*



## CHEMPON DYES (P) LTD.

# 3, Maanasa, 145, St. Mary's Road, Abiramapuram, Chennai - 600 018. India

Tel.: +91-44-4310388, 4310242; Fax: +91-44-432-0816

Website: [www.chempon.com](http://www.chempon.com); E-mail: [dyes@chempon.com](mailto:dyes@chempon.com)

**TABLE 3: LIMITING OXYGEN INDEX AND SMOKE INDEX OF COATING COMPOSITIONS**

Composition	Oxygen Index	Smoke Index
1.	94.4	4.05
2.	92.2	4.22
3.	96.8	3.95
4.	100.0	3.76
Uncoated	42.7	71.53

**Flammability**

Flammability was studied using the Limiting Oxygen Index (LOI) test as per ASTM D 2863<sup>(13)</sup>. LOI is the value at which the material just supports flaming combustion. The minimum concentration of oxygen in a mixture of oxygen and nitrogen flowing upward in a test column that will just support combustion is determined under equilibrium conditions of candle like burning. The limiting oxygen index of a material is determined as:

$$LOI = 100 \times \frac{[O_2]}{[O_2] + [N_2]} \quad [1]$$

PVC sheet of size 52 x140 x5 mm was used to determine the limited oxygen index. The specimen was clamped in the specimen holder and mounted vertically in the center of the test column. A tentative initial oxygen concentration was selected to start the set of experiments. The entire top of the specimen was ignited with the help of propane gas. Ignition flame was then removed and the timer started to record the duration of specimen burning. The objective of the test was to determine the minimum oxygen concentration that just allowed the specimen to continue burning either for three minutes or for at least 75 mm the length of the material. The optimum value of oxygen concentration in the gaseous mixture, flowing through the column is termed as limiting oxygen concentration. The obser-

vations for different compositions are recorded in *Tables 3* and *4*.

**Smoke generation**

Smoke generation from FRSS intumescent coatings and the effect of FRSS additives on smoke suppression in PVC compositions was studied using a dynamic method employing LOI apparatus for the combustion of test specimen<sup>14</sup>. Effect of FRSS additives on smoke suppression was estimated in terms of a smoke suppression index. Smoke generation from coatings was estimated by coating PVC sheets 150 x52 x3 mm with 2-mm thick (dry) film of four different compositions and exposing these sheets to the test. Specimens were mounted vertically and subjected to test, for determining the smoke generation, in the Oxygen Index Apparatus. In the dynamic method, specimens are burnt for three minutes at an oxygen concentration that was 2% higher than the LOI value of that specimen. The smoke evolved is allowed to flow through a duct. A He-Ne laser source and receiver are employed to measure the density of smoke flowing through the duct. The smoke generated by a specimen is represented in terms of 'S' the smoke index, which is derived from the area under the optical density profile i.e. the 'time Vs smoke density' curve<sup>15</sup>. Smoke generation from PVC sheathed cables coated with intumescent coatings was measured in terms of smoke index and the results are reported in *Table 3*. Efficacy of smoke suppressant systems is calculated in terms of the Smoke

Suppression Index (SSI)

$$SSI = \frac{S - S_t}{S} \times 100$$

Where S = area under smoke density curve for sample, and

S<sub>t</sub> = area under the time density curve for the treated

Since the initial weight loss for each sample is identical, A (Area under smoke density curve for sample/weight loss of sample) is replaced by A<sub>t</sub> (Area under the time density curve for the treated sample/weight loss of the sample) in place of S and S<sub>t</sub>. Position (i.e. S > S<sub>t</sub> or A > A<sub>t</sub>) in terms of amount of smoke generated, a negative value represents smoke suppression achieved by use of FRSS in PVC reported in *Table 5*.

**Char formation**

The char formation in different compositions was assessed using the oxygen index apparatus modified by the FRSS additives. The best results in terms of retardance and smoke reduction were observed when exposed to burning conditions. To reduce char in an LOI test, the duration of exposure varies and the oxygen concentration

**TABLE 4: LIMITING OXYGEN INDEX OF PVC COMPOSITIONS**

	Plasticizer								
	Control	DOP							TC
		1	2	3	4	5	6	7	
Phr	-	1.0	1.5	2.0	2.5	3.0	1.0	1.5	2.0
Metal Complex*	-	Mo	Mo	Mo	Mo	Mo	Mo	Mo	Mo
LOI	27.2	27.4	27.7	28.1	28.6	29.2	32.0	32.5	33.0
Metal Complex *	-	Cr	Cr	Cr	Cr	Cr	Cr	Cr	Cr
LOI	27.2	27.6	27.7	27.5	27.8	28.0	31.6	31.8	32.0

\* Metal Based Organic Complex used as FRSS : Mo = Dioxobis (2,4 pentanediono) and Cr = Tris (2,4 pentanediono) chro

**TABLE 5: SMOKE SUPPRESSION INDEX OF PVC COMPOSITIONS**

	Plasticizer										
	Control	DOP					TCP				
		1	2	3	4	5	6	7	8	9	10
Phr	-	1.0	1.5	2.0	2.5	3.0	1.0	1.5	2.0	2.5	3.0
Metal Complex*	-	Mo	Mo	Mo	Mo	Mo	Mo	Mo	Mo	Mo	Mo
SSI	-	27.27	28.88	30.90	48.18	57.57	43.00	46.99	52.98	56.30	58.08
Metal Complex*	-	Cr	Cr	Cr	Cr	Cr	Cr	Cr	Cr	Cr	Cr
SSI	-	19.19	24.85	50.30	32.52	9.09	24.59	25.80	26.37	48.57	58.75

\* Metal based organic complex used as FRSS : Mo = Dioxobis (2,4 pentanediono molybdenum (VI)) and Cr = Tris (2,4 pentanediono) chromium (III)

fire and allowed to naturally. Obs made during the are recorded in

**Results and discussion**

The main cons fire and smoke intumescent coat study are amide phosphates propanediono co transition metals plexes used in t study were pro complexes of m and chromium.

was 2% higher than the LOI value for that specimen. A wire gauge was placed about 15 mm below the specimen for collection of char. The specimen and char were carefully weighed before and after burning. The backbone char percentage (BC%) was computed as described by Kroenke<sup>16</sup> and the results reported in Table 6.

An apparatus 1200 (H) x 300 (W) x 450 (D) mm having an open front and two clamps to hold the cable specimen in vertical position was employed to study the char formation on cables exposed to fire conditions. In accordance with IEC-332<sup>17</sup> a bunsen burner regulated to give a 125-mm long flame having a blue cone of 40 mm was used as the ignition source. The burner was applied at 45° to the vertically held specimen for 'T' seconds. 'T' is calculated as

$$T = 60 + W/25 \quad [3]$$

Where, W is the weight in grams of the 600-mm long cable specimen. After removal of the ignition flame the period of burning, if any, is recorded in seconds. Later, the specimen surface is wiped clean and the distance of charred or fire affected portion of PVC is measured (Table 7).

In this test we have studied the behaviour of a single cable however, the

behaviour of a group of cables may be quite different from it. Thus the flame propagation characteristics of a number of cables laid in a single layer were evaluated employing IEEE-383<sup>18</sup> method. Seven, PVC sheathed multistrand aluminium conductor cables of 17 mm outer diameter coated to an average thickness of 1.5 mm were arranged in a metallic tray of 2440 (L) x305 (W) wide x 76 (D) mm in a single layer. 610 x 610 mm piece of burlap having a unit weight area of 0.3 kg/m<sup>2</sup> was made into a bundle of 100x100 x 150 mm size and soaked in super F multigrade 20W/40 petrol engine oil so as to retain 160 g of it. The soaked burlap was used as igniter and held at the center and in front of the cable tray at a point 600-mm above the bottom of the cable tray. The burlap was set on

combination is exposed to fire phosphate decomposes to phosphoric acid, which acts as a dehydrating agent. The polyol is dehydrated acid, forming a large amount of carbonaceous char that produces a protective char barrier to protect the cable. Amide gives off non-flammable gases causing the foamable carbonaceous char to produce a honeycomb blanket of char, a highly effective insulation. On softening forms an expanded char over the carbonaceous char, preventing the escape of gases produced during burning. MBO complexes are chelated transition metals having an organic ligand that can cleave at an elevated temperature thereby releasing the metal in a reactive state for its action as an additive. These additives limit the char length, thereby restricting

**TABLE 6: CHAR FORMATION OF FRSS MODIFIED PVC COMPOSITIONS**

	Plasticizer										
	Control	DOP					TCP				
		1	2	3	4	5	6	7	8	9	
Phr	-	1.0	1.5	2.0	2.5	3.0	1.0	1.5	2.0	2.5	
Metal Complex*	-	Mo	Mo	Mo	Mo	Mo	Mo	Mo	Mo	Mo	
BC%	34.79	71.25	74.31	78.07	79.11	61.38	62.94	71.40	73.87	76.4	
Metal Complex*	-	Cr	Cr	Cr	Cr	Cr	Cr	Cr	Cr	Cr	
BC%	34.79	51.93	58.62	61.43	60.00	56.27	49.13	51.30	53.58	55.7	

\* Metal Based Organic complex used as FRSS : Mo = Dioxobis (2,4 pentanediono) molybdenum (VI) and Cr = Tris (2,4 pentanediono) chromium (III)

**TABLE 7: PERFORMANCE OF COATING AS PER IEC-332**

Composition No.	Cable diameter (mm)	Weight of specimen (gm)	Exposure time (second)	Continuation of flame after exposure (second)	Charred length mm	Affected PVC portion (mm)
1.	20	442.8	78	0	80	Nil
2.	20	448.2	78	0	78	Nil
3.	20	440.9	78	0	74	Nil
4.	20	447.0	78	0	70	Nil
Uncoated	17	350.2	74	23	178	168

lecular cyclization and hence resulting in lower amount of smoke generation. Their efficacy in suppressing smoke generation could be ascribed to their better miscibility with the polymer. Essentially all of the effective metal based smoke suppressants appear to work in solid phase resulting in increased char formation and thus reduced flammability of the polymer. They also interfere with the normal degradation pattern of PVC<sup>19</sup>. Smoke retardants act through increase in char formation as well as through change in pyrolyzates. This results in slower burning and lower smoke generation.

Since the maximum effectiveness of MBO additives in suppressing smoke emission is obtained at relatively small concentrations it was decided to use pentanediono complexes of molybdenum (VI) and chromium (III) up to 3 parts per hundred parts of resin (phr) loading. Samples were prepared as described earlier, using both the phthalate (dioctyl phthalate- DOP) and the phosphate (tri cresyl phosphate - TCP) plasticizer at 50 phr. A sample containing 0.0 phr of the additives was used as the control or the reference sample for computation of smoke suppression index (SSI). Experimental results indicate that the oxygen index of coatings varied between 92 and 100. This indicates that these coatings are not likely to catch fire even under the worst conditions, where as the un-coated rigid PVC sheet used as the substrate was found to be a fairly fire retardant one (Table 3). Plasticized PVC sheets modified with FRSS additives also proved to be fairly flame retardant particularly

when phosphate plasticizer was used (Table 4).

Smoke generation was found to have reduced very significantly with all the formulations under study (Table 3). For a similar exposure period of three minutes the smoke index was encountered to be sharply decreased from 71.53 for the substrate to as low as 3.76 in case of formulation 4. When the MBO used was dioxobis (2,4 pentanediono) molybdenum (VI) constants decrease in smoke generation i.e. an increase in SSI from 27.27 to 57.57 was observed in case of DOP plasticizer. Comparatively higher values of smoke suppression are observed in the case of samples plasticized with the phosphate plasticizer. The SSI increased from 43.00 to 58.08 when the loading of the molybdenum complex was increased from 1 to 3 phr (Table 5). Samples modified by 1-3 phr of Tris (2,4 pentanediono) chromium (III) showed an overall trend of smoke suppression vis a vis quantity of the complex used for both the DOP and TCP plasticized samples. Phthalate plasticized samples reveal maximum smoke

suppression at 2 phr. An increase in loading of the complex was also observed. In the case of plasticized sample, maximum smoke suppression was observed at 3.0 phr level. In the case of the above behavior, the complex interferes with the degradation (from TCP) to resin char formation while the plasticizer results in its own right.

Quantity of char formed increases with the amount of the additive upto

it shows a decrease in the degradation process. Lower aliphatics instead of aromatic seems to be the reason for the formation of char when more of char when more of the additive was used. The formation in DOP plasticized samples is modified by the chromic complex (through crosslinking) which has been replaced by a phosphate complex. Phosphates resulting in an increase in char formation and hydrocarbon emission. An increase in char formation with an increase in quantity of MBO used was observed when phosphate plasticizer was used. This is basically because they have a tendency to

Cable specimen of different compositions tested at different thickness tested as per IEC-332 were found to extinguish the flame. The source was removed and the specimen was removed. This was noted in the case of the uncoated and only charring was observed. The coating up to 80 phr was the source of flame was found unaffected during burning. On the other

**TABLE 8: FIRE PERFORMANCE AS PER IEEE-383**

Composition No.	Affected charred length (mm) of different specimens				
	C <sub>1</sub>	C <sub>2</sub>	C <sub>3</sub>	C <sub>4</sub>	C <sub>5</sub>
1.	75	338	370	302	314
2.	40	305	350	336	340
3.	24	310	340	285	285
4.	Nil	210	270	165	170
Uncoated	404	862	1050	985	900

# polyester polyols

Widest Range. Customised Specs. Finest Quality.  
for specialised PU applications

Footwear

Flexible Foam

Coatings & Paints

Elastomers

Adhesives

**EXPANDED  
INCORPORATION**

C-44/1 TTC Area, MIDC, Pawane Village, Vashi, Navi Mumbai 400 705. Tel. 768 0303/0404, 767 0838/39 Fax. 767 1065 eMail. expanded.in

specimen charring was noted up to 178 mm and PVC portion was totally damaged up to 168 mm under the same exposure conditions of IEC-332 (Table 7).

It is evident from Table 8 that the coating prevented the propagation of flame in different cables when they were arranged in a single layer. It is also noted that coated cable did not burn out to the total height of the cable tray above the flame source. Maximum affected charred length was observed in cables number C<sub>3</sub>, C<sub>4</sub> and C<sub>5</sub> which were centrally located in cable tray and were in direct touch of heat source. Minimum charred length was observed in adjacent cables number C<sub>1</sub>, C<sub>2</sub>, C<sub>6</sub> and C<sub>7</sub> which show that coated specimens were retarding flame spread. On the other hand, it can be explained that the charred length was due to direct touch of burlap flame that was measured about 350 to 400 mm height. All the uncoated cable specimens were found to be severely affected and the PVC insulation was found to be totally damaged, however, in case of the coated specimens only exfoliation of coating was observed and no bare wires were seen even after the total burning of burlap igniter.

## Conclusions

A few flame retardant smoke suppressant intumescent coatings based on indigenously available chemicals were developed. The coatings applied on PVC sheathed electric cables were found to be quite effective in reducing the flammability and smoke generation from the electrical cables. Fire retardant behaviour (estimated in terms of LOI) and smoke suppressant behavior (estimated as smoke suppression index - SSI) of PVC modified with metal based

organic complexes reveal that the additive acts as a FRSS with both the phthalate and the phosphate plasticizers. However, their overall performance is superior in the case of the phosphate plasticized PVC. The char formation by PVC sheets, an indicator of FRSS behaviour, was also found to have increased. However, charring, an indicator of extent of burning for the coated cables was found to have decreased significantly when the intumescent coating, reported herein, was used.

## References

1. Lawson D.F., in *Flame Retardant polymeric Materials*, M. Lewin, S.M. Altas and E.M. Pearce (eds), Vol.3 Plenum Press, N.Y., (1982) 39-95.
2. Sharma Sunil K., Metal Based Organic Complexes as Fire Retardant Smoke Suppressant for Polyvinyl chloride, *Research and Industry*, **38**, (1993), 268-272.
3. Sharma Sunil K. and S.K. Srivastava, Smoke Suppression in Poly (vinyl chloride); *Fire Safety Science*, **4** (2), (1995), 38-51.
4. Loyns J W, *The chemistry and uses of fire retardants* (New York, Wiley Interscience), (1970), 256-272
5. Baker D S, Flame retarding wood and timber products, *Chem Ind* **2**, (1977), 74-79.
6. Dua A C, Fire retardant paints - Effects of various additives in their formulations, *Paint India*, **32** (7), (1982), 3-5.
7. Bhatnagar V M & Vergnaud J M, Fire retardant paints, *Paint India*, **33** (7), (1983), 15-17.
8. Jain J P, Saxena N K & Singh Ilam, Fire retardant surface coating for cellulosic materials, *Research and Industry*, **30** (1), (1985), 20-24
9. Saxena N K, & Gupta D R, Development and evaluation of fire retardant coatings, *Fire Technol*, **26** (4), (1990), 311-318.
10. Saxena N K, Sharma D R, "Fire protection products by coating", *Sci.*, **21** (2). (1990), 105-108.
11. Smith, T., Flame resistant paints and coatings, *Fire Technol*, **11** (4), (1977), 205-210.
12. Sharma Sunil K., *Studies on Fire Retardant Products of Some Polymeric Materials*, CCS University, Meerut, India, (1991).
13. ASTM D 2863 Standard Test Method for Determining the Minimum Temperature to Support Combustion of Plastics (Oxygen Index), American Society for Testing and Materials, Philadelphia, PA 19103, (1990).
14. Sharma Sunil K., N.K. and Srivastava, Effectiveness of Smoke Suppressant Using a He-Ne Laser, *Fire and Materials*, **19** (1), 7-11.
15. Singh M P and Singh N, Smoke Density Versus White Light Scattering (Optical Density, *Fire Technol*, **19** (1), 7-11.
16. Kroenke W.J., M. J. and M. J., Smoke for Poly (vinyl chloride), *Fire Technol*, **26**, 1167-1171.
17. IEC-332 "Tests on Fire Conditions. Part 1: Vertical Insulation Test", International Electrotechnical Commission, Geneva Suisse, (1975).
18. ANSI/IEEE-383 'Standard Test Method of Class IE Electrical Connections: Generating Stations, Industrial and Electrical Connections', (1974).
19. Saxena N.K., Saxena T.P., Fire Protection by Coating, *Fire Technol*, **11** (4), 233-240.
20. Lattimer R.P., The protective role of Fire Retardant Coatings (vinyl chloride), *Fire Technol*, **11** (4), 1190-1210, (1977).

## Information on your new products, equipments and processes

You are invited to submit information about new products, equipments, and processes, developed and offered by you to the Indian market. Such writeups are entertained for publication without any cost.

Don't miss this opportunity to promote your new products.